

IGD-TP 4th Exchange Forum

WG3: New Waste Type, in collaboration with SNETP

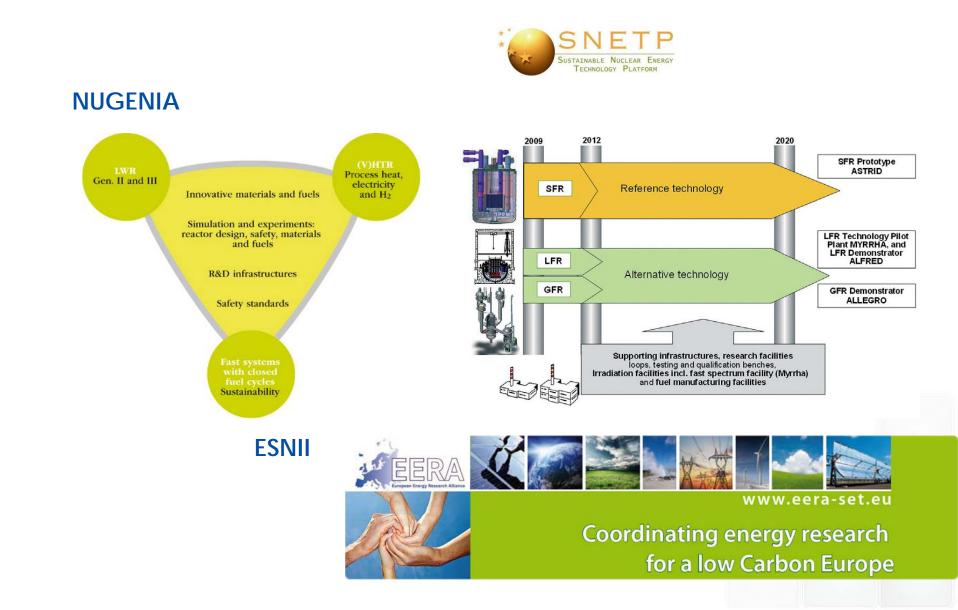
Long-term Behavior of Waste Forms from Gen IV Reactors towards Geological Disposal

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Prague, October 29-30, 2013

SNETP pillars and IV Generation Technologies





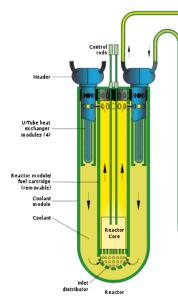
LFR – Lead Fast Reactor Demostrator - ALFRED





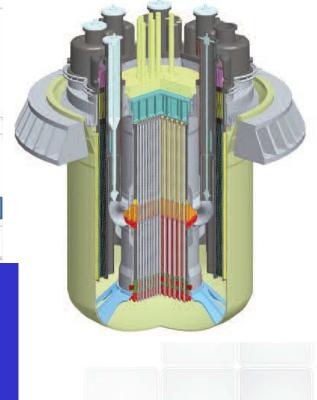






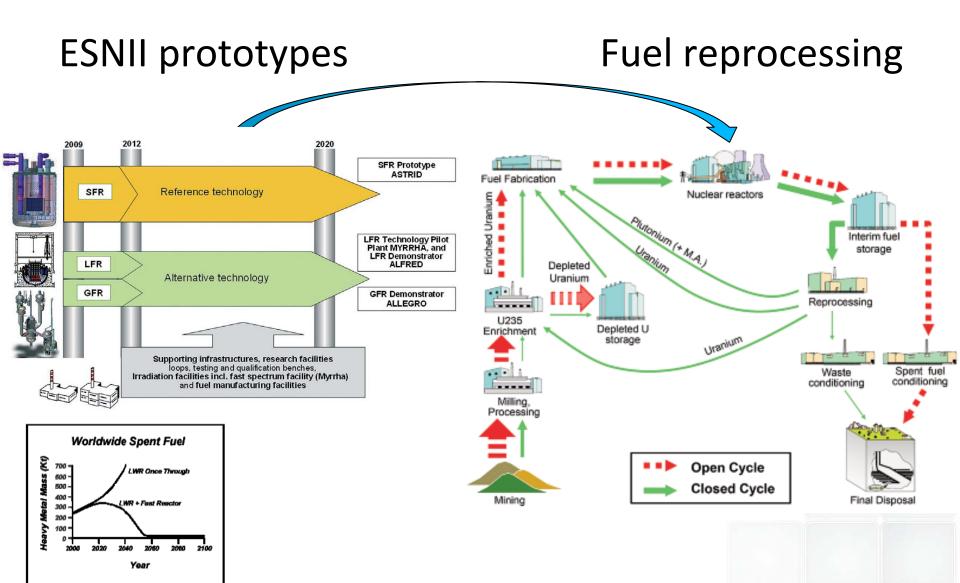
Generator Generator Turbine Compressor Compressor

Characteristics • Pb or Pb/Bi coolant • 550° C to 800° C outlet temperature • Fast Spectrum • Multi-TRU recycle • 50–1200 MWe • 15–30 year core life



Congresso Annuale 2012 SISN (Societ Fiorentino

Gen IV reactors spent fuel reprocessing



AGENZIA NAZIONALE

PER LE NUOVE TECNOLOGIE, L'ENERGIA E LO SVILUPPO ECONOMICO SOSTENIBILE

Fuel Cycle Innovations





- Advanced aqueous or pyroprocess recycle technology with intrinsic nonproliferation features
 - Produces co-mixed Pu+MA feedstock (no separated Pu)
 - Incomplete fission product separation
- Fuel Cycle Support Facilities Options
 - Co-located
 - Regional
- Multiple Fuel Options
 - Fuel types TRU/U/Oxide, TRU/U/Nitride, TRU/U/Alloy, TRU Alloy, MA Alloy
 - Remote refabrication technology depends on fuel type
 - Simplified pelletization (oxide, nitride)
 - Vibrocompaction (oxide, nitride)
 - Injection casting (metal)

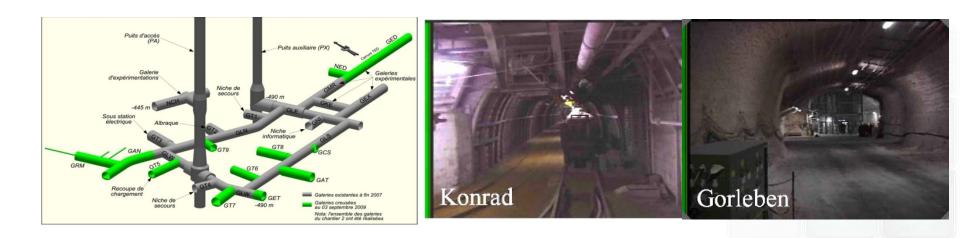
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New waste form requirements from Gen IV reactors reprocessing to geological disposal

- Burn most of TRU / Minor Actinides
- Reducing FP as much as possible
- Non proliferation of Pu
- Stability of chemical and physical form
- Easy conditioning and storage
- Safe and economic geological disposal
- Average depth around m.500 and not m. 4000

GEOLOGIC PROFILE Surficial Sand Feet Ground Level Dewey Lake Redbeds F40 Rustler Formation B50 1000 Comparison Comparis

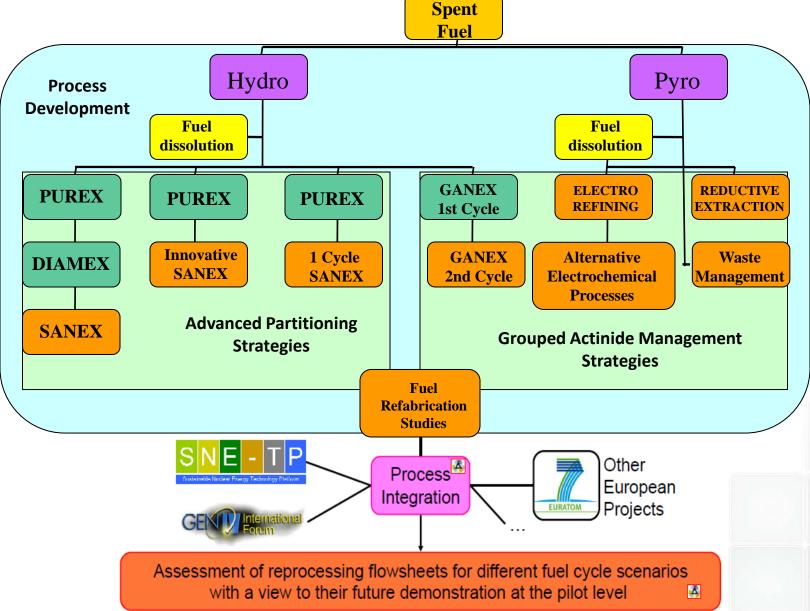
Salado Formation 2000 2150 Waste Repository Level Evaporites (Salt) Sea Level 3000 3400 Sea Level Castile Formation 4000 Salt and Anhydrite Bell Canyon Formation 4500 Salt and Anhydrite





Treatment of spent fuel







The main reprocessing methods presently used:

Co-extraction

New extraction

• Uranium extraction (UREX)

• Electrochemical processing



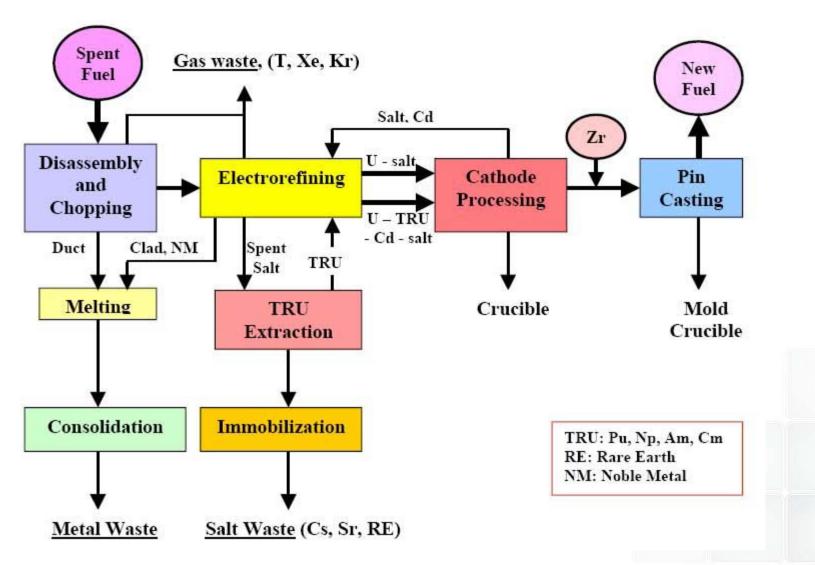


Electrochemical processes



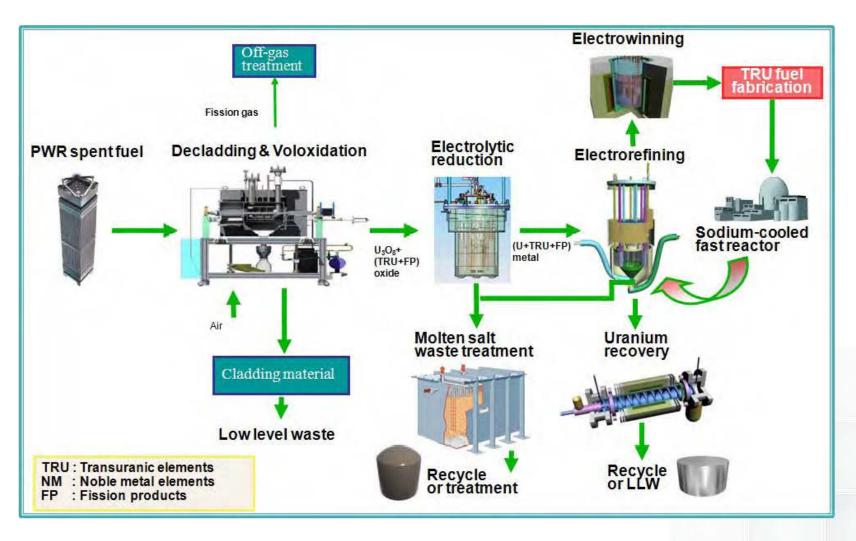


Block diagram of the pyrochemical process for metal fuel





Schematic illustration of the pyrochemical process for oxide fuel





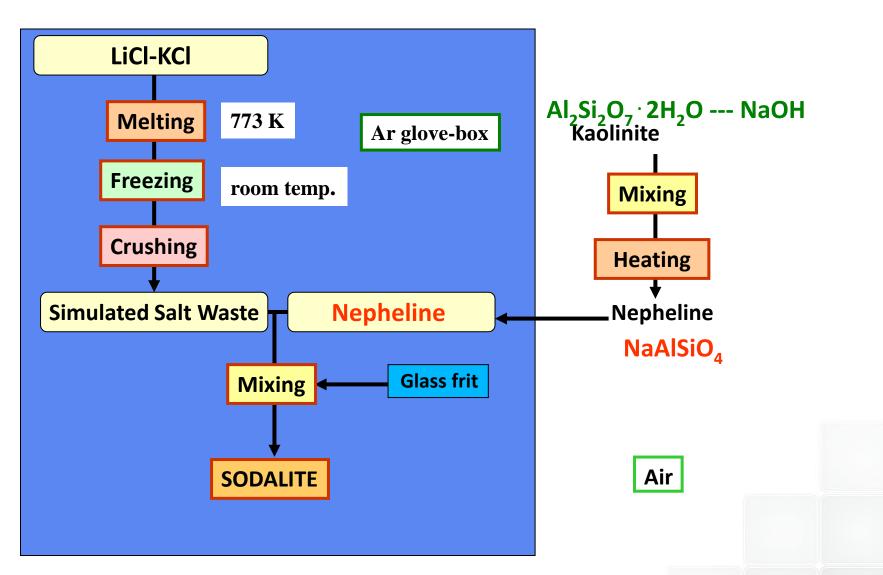


Conditioning of chloride salt wastes from pyroprocesses with different matrices:

SODALITE SAP MURATAITE-PYROCHLORE



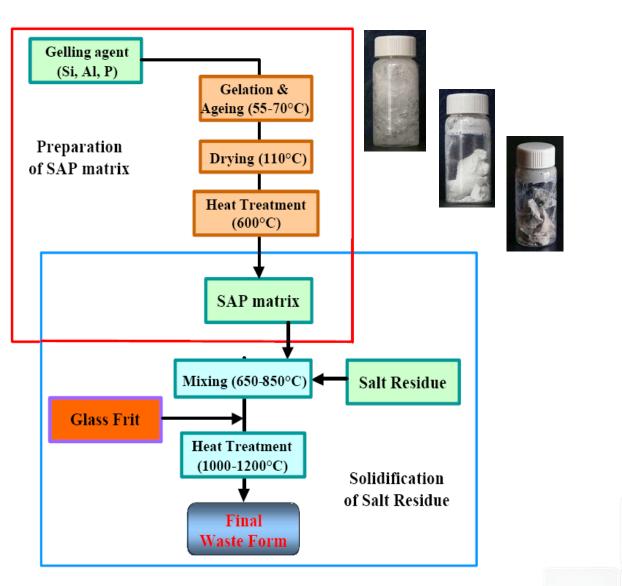
SODALITE matrix



ENER

Outline of sodalite synthesis from kaolinite through nepheline

SAP matrix

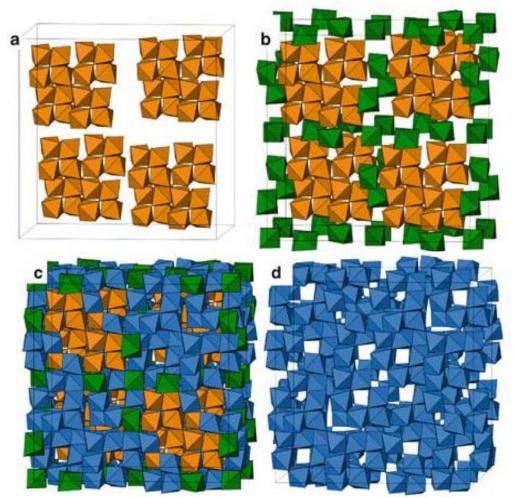


ENER

Outline of SAP synthesis by a conventional sol-gel process



MURATAITE matrix



Octahedral framework in the structure of Mu-5: arrangement of pyrochlore clustera formed by corner sharing of *Ti*1O6 and *Ti*4O6 octahedra (a); linkage of pyrochlore clusters by Ti3O6 octahedra (b); whole framework as combination of linked pyrochlore clusters and murataite-like framework formed by Ti2O6 and Ti5O6 octahedra ©; murataite-like framework (d)



Evaluation of long-term behaviour of conditioned salt wastes

- Characterization of the final waste forms
- Static leaching tests up to 150 days
- Assessment of the main parameters which influence fission products release (pH, temperature, contact time)
- Determination of the interactions between host matrix and individual fission elements



Characterization of the final waste forms

Density measurements; Thermogravimetric analysis; SEM-EDS; Optical Microscopy; FTIR; XRD



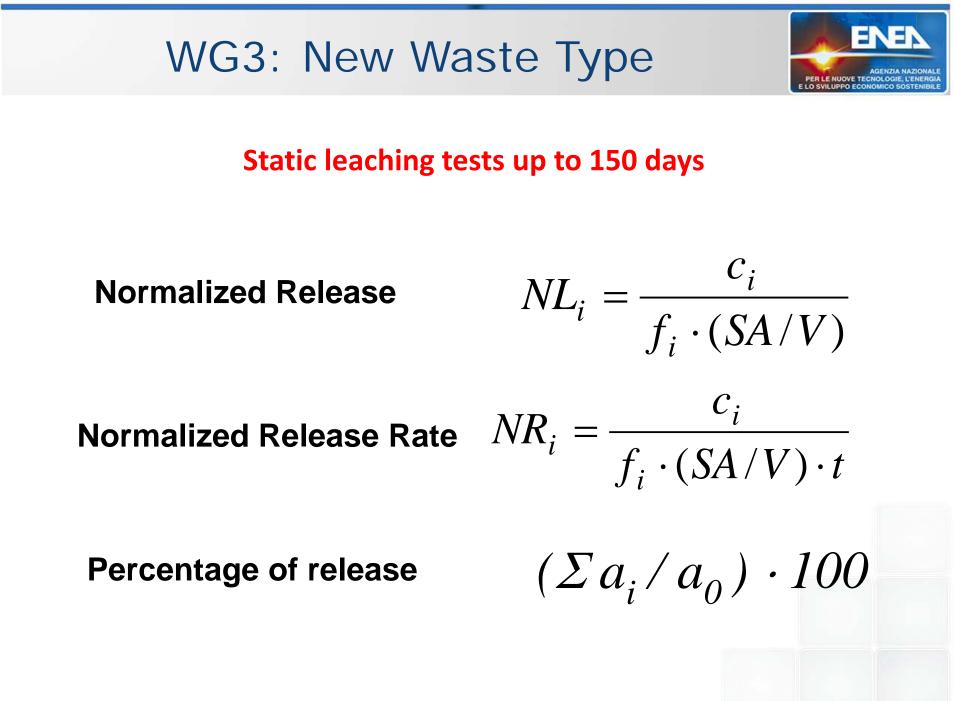




Static leaching tests up to 150 days

Reference Method

C 1285-02 - Standard Method for Determining Chemical Durability of Nuclear, Hazardous, and Mixed Waste Glasses and Multiphase Glass Ceramics: the Product Consistency Test (PCT)





Assessment of the main parameters which influence fission products release (pH, temperature, contact time)

pH: acid; neutral; alkaline

Temperature: 23C; 90C

Contact time: 1; 3; 7 days





Determination of the interactions between host matrix and individual fission elements

Single fission elements incorporated in the host matrix with characterization of the mineral phases formed:

Alkaline metal: Cesium

Alkaline-earth metal: **Strontium**

Rare earth metal:

Neodimium



CONCLUSIONS

The main scope of this proposal is a comparison among these promising matrices for conditioning of chloride salt wastes, with a particular attention to the incorporation mechanisms and to their durability in the long-term, in order to comply with geological disposal





Thanks for your attention

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